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Parallel solution phase synthesis of a library of amino acid derived 2-arylamino-[1,3,4]-oxadiazoles

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ABSTRACT

A mild method for the synthesis of peptidomimetic 2-arylamino 5-substituted 1,3,4-oxadiazoles from Boc-protected α -amino acid derived hydrazides has been developed, and applied in a parallel solutionphase synthesis. The optimized reaction conditions involve a one-pot reaction of Boc-protected amino acid hydrazides with arylisothiocyanates in the presence of either Hg(II) chloride, Mukaiyama's reagent (2-chloro-N-methylpyridinium iodide) or polymer supported Mukaiyama's reagent, with triethylamine in dichloromethane at ambient temperature. The 1,3,4-oxadiazole products were obtained in good to excellent yields without any detectable epimerization. The reactions proceed via initial formation of thiosemicarbazides, followed by dehydrothiolative cyclization to the 1,3,4-oxadiazoles.

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1,3,4-Oxadiazole derived compounds are known to display a wide range of biological activities including anti-inflammatory, analgesic, antibacterial, fungicidal, antimicrobial, hypoglycemic, antimalarial, antitubercular, antidepressant and other activities.¹ 1,3,4-Oxadiazoles can also serve as bioisosteric replacements of amide and ester functionalities, $²$ $²$ $²$ and can participate in hydrogen</sup> bonding interactions as hydrogen bond acceptors. As such, their peptidomimetic ability has been explored^{[3](#page-2-0)} and reported in the development of Phe-Gly mimetics of dermorphin (Tyr-D-Ala-Phe-Gly-Tyr-Pro-Ser-NH2) and Substance P (Arg-Pro-Lys-Pro-Gln-Gln-Phe-Phe-Gly-Leu-Met-NH₂).^{[4](#page-2-0)} In addition to their utility as bioactive molecules, 1,3,4-oxadiazoles are useful intermediates for organic synthesis,^{[5](#page-2-0)} particularly as electron-deficient azadienes in inverse electron demand Diels-Alder reactions.^{[6](#page-2-0)}

As part of our interest in the synthesis of peptidomimetic heterocycle-peptide hybrid molecules, 7 we now report an approach to the synthesis of amino acid derived 2-arylamino-[1,3,4]-oxadiazoles. Most of the synthetic methods available for the construction of 1,3,4-oxadiazoles involve rather forcing conditions and/or high temperatures that are unsuitable for use with amino acids or peptides.⁸ Thus, we aimed to develop a reliable method, mild enough to avoid epimerization problems, and that would be compatible with orthogonal protecting groups. In addition, it was desirable that the approach be applicable to parallel solutionphase synthesis and, potentially suitable for combinatorial solidphase library production. We envisaged that an approach based upon the dehydrothiolative cyclization of α -amino acid or peptide derived acylthiosemicarbazides would generate the corresponding 1,3,4-oxadiazoles. Although such a strategy has not been previ-

* Corresponding author. Tel./fax: +1 416 978 5059. E-mail address: rbatey@chem.utoronto.ca (R. A. Batey). ously applied to amino acid derived 1,3,4-oxadiazoles, dehydrothiolation of simple acylthiosemicarbazides to give 1,3,4-oxadiazoles has been reported using a variety of reagents including lead oxide, mercury oxide, mercury acetate, copper sulfate, bromine, $I_2/NaOH$ and DCC. 9 Moreover, the strategy we envisaged is similar to those we have previously employed for the synthesis of α -amino acid and peptide derived 5-aminotetrazoles and 2-iminohydantoins, $7,10$ involving activation of thiourea intermediates.

Several Boc-protected amino acids (Phe, Leu, Ala) were first converted to the corresponding hydrazides 1 following a literature procedure using EDCI activation.¹¹ Reaction of the hydrazides 1 with arylisothiocyanates in $CH₂Cl₂$ for 20 min at room temperature provided the corresponding Boc-thiosemicarbazides 2a–h quanti-tatively, without the need for chromatographic purification^{[12](#page-3-0)} ([Ta](#page-1-0)[ble 1](#page-1-0)). Deprotection of 2 using 20% TFA/CH₂Cl₂ solution and treatment of the resulting thiosemicarbazides with $HgCl₂$ in the presence of triethylamine in acetonitrile for 12 h at room temperature gave the desired oxadiazole products 3a–h in 78–85% isolated yields [\(Table 1\)](#page-1-0).

The yields of the oxadiazoles were improved, and the purification protocol simplified, if the Boc group was not removed prior to the dehydrothiolative cyclization. Thus, reaction of Boc-protected hydrazides 1 with arylisothiocyanates gave the corresponding thiosemicarbazides $2¹²$ $2¹²$ $2¹²$ which on treatment with triethylamine and $HgCl₂$ led to the Boc-protected 1,3,4-oxadiazoles 4 [\(Table](#page-1-0) 2).^{13,14} This method was suitable for parallel solution-phase synthesis, using semi-automated flash chromatographic purification. Twenty-four examples were prepared in parallel fashion starting from Boc-Ala-OH, Boc-Leu-OH and Boc-Phe-OH in combination with eight different arylisothiocyanates [\(Table 2](#page-1-0)). The products 4 were obtained in good to excellent yields, after reaction for 12 h at ambient temperature.

Table 1

Synthesis of amino acid derived acylthiosemicarbazides 2 and 5-substituted 2-arylamino 1,3,4-oxadiazoles 3^a

^a Reagents: (a) ArNCS (1.0 equiv), CH₂Cl₂, rt, 20 min. (b) 20% TFA/CH₂Cl₂, rt, 1.5 h. (c) HgCl₂ (1.1 equiv), Et₃N (1.1 equiv), MeCN, rt, 12 h.

Isolated yields obtained after column chromatographic purification (1% Et₃N in EtOAc:Hexanes 1:4).

The structure of the product oxadiazole 4Ce was confirmed by X-ray crystallography (Fig. 1). 15 The solid-state conformation of **4Ce** revealed that the oxadiazole and bromophenyl rings were coplanar, maximizing conjugation of the anilino-N lone-pair.

A selection of eight of the N-Boc-protected compounds 4 were treated with 50% TFA/CH₂Cl₂ solution to give the deprotected

Table 2

Parallel synthesis of N-Boc-protected 5-substituted 2-arylamino 1,3,4-oxadiazoles 4 using HgCl₂ and $Et₂N²$

Boc.	R ¹ H NH ₂ н O	(a), (b)	R ¹ Boc. Ν Ĥ $N-N$	R^2 Н
Entry	1 Product	R ¹	R ²	4 Yield ^b %
$\mathbf{1}$	4Aa	Me	H	85
$\overline{2}$	4Ab	Me	$2-Me$	88
3	4Ac	Me	3-Me	86
4	4Ad	Me	4-Me	87
5	4Ae	Me	$2-Pr$	97
6	4Af	Me	$3-Br$	99
$\overline{7}$	4Ag	Me	$4-NO2$	80
8	4Ah	Me	$3,4,5-(MeO)_3$	98
9	4Ba	$i-Pr$	H	92
10	4Bb	$i-Pr$	$2-Me$	88
11	4Bc	$i-Pr$	3-Me	80
12	4Bd	$i-Pr$	4-Me	89
13	4 _{Be}	$i-Pr$	$2-Pr$	98
14	4Bf	$i-Pr$	$3-Br$	99
15	4Bg	$i-Pr$	$4-NO2$	85
16	4Bh	$i-Pr$	$3,4,5-(MeO)_3$	96
17	4Ca	Bn	H	89
18	4Cb	Bn	$2-Me$	89
19	4Cc	Bn	3-Me	80
20	4Cd	Bn	4-Me	87
21	4Ce	Bn	$2-Pr$	99
22	4Cf	Bn	$3-Br$	99
23	4Cg	Bn	$4-NO2$	75
24	4Ch	Bn	$3,4,5-(MeO)3$	97

Reagents: (a) ArNCS (1.0 equiv), CH_2Cl_2 , rt, 20 min. (b) $HgCl_2$ (1.1 equiv), Et_3N (1.1 equiv), MeCN, rt, 12 h.

b Isolated yields obtained after column chromatographic purification (EtOAc:Hexanes 1:4).

Figure 1. X-ray crystal structure of the 1,3,4-oxadiazole 4Ce. Thermal ellipsoids are shown at 30% probability.

compounds 3 as their TFA salts. The products were obtained in excellent yields and high purities without the need for chromatographic separation (Table 3). The yields and purities of the compounds 3a–f,h obtained in this manner (i.e., synthesized via 4) were comparable to the results for the syntheses of these compounds where N-Boc deprotection preceded oxadiazole formation (i.e., Table 1). In general, however, synthesis and storage via the N-Boc-protected compounds 4 is preferable due to the greater long-term stability of 4 and the TFA salt of 3, compared to the free-base form of 3.

Proline derived 1,3,4-oxadiazoles 6 can also be synthesized using these two approaches [\(Table 4\)](#page-2-0). In the first approach, initial formation of the Pro-derived thiosemicarbazide was followed by Boc-deprotection using 20% TFA/CH₂Cl₂, and then immediate cyclization to give **6a–h**. This approach had the advantage of not requiring isolation of any of the intermediates. However, the crude products $6a-h$ were obtained with only 80–87% purity,¹⁶ necessitating flash chromatographic purification of 6. In the second approach, the intermediate thiosemicarbazides were first cyclized to give $5a-h$, and subsequently deprotected using 50% TFA/CH₂Cl₂

Table 3 Boc group deprotection of 4 into 3

^a Yield of the unchromatographed product.

^b Determined by HPLC/LCMS with the UV absorption measured at 254 nm.

Table 4

Synthesis of proline derived 1,3,4-oxadiazoles 5 and 6^a

Reagents: (a) ArNCS (1.0 equiv), CH_2Cl_2 , rt, 20 min, (b) 20% TFA/CH₂Cl₂, rt, 1.5 h, (c) $HgCl_2$ (1.1 equiv), Et₃N (1.1 equiv), MeCN, rt, 12 h, (d) 50% TFA/CH₂Cl₂, rt, 30 min. b Isolated yields obtained after column chromatographic purification based on the starting Boc-protected hydrazide 1d.

Isolated yields of 6 produced using $(a)-(c)$.

 d Isolated yields of 6 prepared by deprotection of 5 were quantitative.

to give 6a–h. The initial Boc-protected products 5 were isolated in good yields. Again, column chromatographic purification following the cyclization step was required, since the crude products 5 had 85–93% purity.^{[16](#page-3-0)} Deprotection of $5a-h$ occurred in quantitative yields to give **6a–h** as the TFA salts in excellent purities ($\geq 98\%$, as confirmed by $^1\mathrm{H}$ NMR analysis) without the need for chromatographic purification. Overall, although both approaches required a single chromatographic purification, purification of 5 was more straightforward than for 6, and the overall yields of 6 were higher via 5. In addition, the Boc-protected compounds 5 were suitable for long-term storage.

We were also interested in developing conditions using alternative dehydrothiolating agents for the reaction, to avoid the use of $HgCl₂$. As a first step, the formation of 1,3,4-oxadiazole **4Ce** from 1a was compared using three different dehydrothiolating agents: HgCl2, Mukaiyama's reagent (2-chloro-N-methylpyridinium iodide, 7^{17} 7^{17} 7^{17}), and its polymer-supported equivalent $8^{17,18}$ $8^{17,18}$ $8^{17,18}$ (Table 5). Each experiment was conducted at ambient temperature for 16 h, affording 4Ce in excellent yields (92–95%) with excellent purities (98–99%). For compound **4Ce**, synthesized using either HgCl₂ or Mukaiyama's reagent, purification by column chromatography was required. For compound $4Ce$ synthesized by the HgCl₂ protocol, NMR analysis of the crude sample showed excellent purity. However, ICP AES analysis revealed the presence of ${\sim}200$ ppm of

Table 5

Comparative study of dehydrothiolating agents used for the formation of 4Ce

^a Isolated yield after column chromatography.

Hg in the crude solid sample after aqueous work-up, while the Hg level of the same sample after flash chromatography was not detectable by ICP AES. The use of Mukaiyama's reagent is of course advantageous due to its lower toxicity; however, the products obtained using this method must also be purified by column chromatography to remove the organic by-products. The use of the polymer-supported CMPI equivalent 8 avoids the toxicity/environmental problems associated with the use of mercury salts, while requiring only a simple filtration upon completion of the reaction. Thus, in the case of the compound 4Ce prepared using solid supported reagent 8, NMR analysis of the crude sample showed excellent purity, and column chromatographic purification was not required. Another operational simplification permitted by the use of the polymer-supported reagent 8 is that all of the reagents can be added at the same time (i.e., without the pre-treatment of the hydrazide with the arylisothiocyanate). Thus, stirring of 1a, 2-bromophenyl isothiocyanate (1 equiv), 7 (5 equiv) and triethylamine (5 equiv) for 16 h gave **4Ce** in 94% yield, compared to a 95% yield (Table 5) obtained for the two-stage addition. Finally, a possible concern for α -amino acid derived compounds is whether stereochemical integrity is retained in the products. Chiral HPLC determination (Chiralcel OD column) of independently synthesized (R) -4Ce and (S) -4Ce verified that epimerization was not observed $(\leq 0.01\%)$ under the reaction conditions.

In conclusion, a convenient method for the synthesis of 2-arylamino 5-substituted 1,3,4-oxadiazoles from Boc-protected amino acid derived hydrazides has been developed. Further studies on the synthesis and applications of 1,3,4-oxadiazoles as peptidomimetic building blocks will be reported in due course.

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References and notes

- 1. (a) Amir, M.; Shikha, K. Eur. J. Med. Chem. 2004, 39, 535–545; (b) Adelstein, G. W. J. Med. Chem. 1973, 16, 309–312; (c) Morton, H. M.; Heuser, D. J.; Joyner, C. T.; Batchelor, J. F.; White, H. L. J. Med. Chem. 1996, 39, 1857–1863; (d) Chavan, V. P.; Sonawane, S. A.; Shingare, M. S.; Karale, B. K. Chem. Heterocycl. Compd. 2006, 42, 625–630; (e) Amir, M.; Kumar, S. Pharmazie 2005, 60, 175–180; (f) Swain, C. J.; Baker, R.; Kneen, C.; Moseley, J.; Saunders, J.; Seward, E. M.; Stevenson, G.; Beer, M.; Stanton, J.; Watling, K. J. Med. Chem. 1991, 34, 140–151.
- Guimaraes, C. R. W.; Boger, D. L.; Jorgensen, W. L. J. Am. Chem. Soc. 2005, 127, 17377–17384 and references cited therein.
- Orlek, B. S.; Blaney, F. E.; Brown, F.; Clark, M. S. G.; Hadley, M. S.; Hatcher, J.; Riley, G. J.; Rosenberg, H. E.; Wadsworth, H. J.; Wyman, P. J. Med. Chem. 1991, 34, 2726–2735.
- 4. Borg, S.; Vollinga, R. C.; Labarre, M.; Payza, K.; Terenius, L.; Luthman, K. J. Med. Chem. 1999, 42, 4331–4342.
- 5. For a review, see: Warrener, R. N. Eur. J. Org. Chem. 2000, 3363–3398.
- 6. For example, see: (a) Ishikawa, H.; Elliott, G. I.; Velcicky, J.; Choi, Y.; Boger, D. L. J. Am. Chem. Soc. 2006, 128, 10596–10612; (b) Elliott, G. I.; Fuchs, J. R.; Blagg, B. S. J.; Ishikawa, H.; Tao, H.; Yuan, Z.-Q.; Boger, D. L. J. Am. Chem. Soc. 2006, 128, 10589–10595; (c) Wilkie, G. D.; Elliott, G. I.; Blagg, B. S. J.; Wolkenberg, S. E.; Soenen, D. R.; Miller, M. M.; Pollack, S.; Boger, D. L. J. Am. Chem. Soc. 2002, 124, 11292–11294; (d) Wolkenberg, S. E.; Boger, D. L. J. Org. Chem. 2002, 67, 7361– 7364.
- 7. (a) Gavrilyuk, J. I.; Evindar, G.; Batey, R. A. J. Comb. Chem. 2006, 8, 237–246; (b) Gavrilyuk, J. I.; Evindar, G.; Batey, R. A. J. Comb. Chem. 2007, 9, 644–651.
- 8. For reviews on 1,3,4-oxadiazoles and leading references on their synthesis, see: (a) Behr, L. C. In The Chemistry of Heterocyclic Compounds; Weissberger, A., Ed.; Wiley: New York, 1962; Vol. 17, pp 263–282; (b) Nesynov, E. P.; Grekov, A. P. Russ. Chem. Rev. (Engl. Transl.) 1964, 33, 508–514; (c) Hetzheim, A.; Möckel, K. Adv. Heterocycl. Chem. 1966, 7, 183-224; (d) Hill, J. In Comprehensive Heterocyclic Chemistry; Potts, K. T., Katritzky, A. R., Rees, C. W., Eds.; Pergamon: Oxford, 1984; Vol. 6, pp 427–446; (e) Hill, J. In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R., Rees, C. W., Scriven, E. F. V., Eds.;

17.

.

Pergamon: Oxford, 1996; pp 267–287; (f) Hetzheim, A. In Houben-Weyl Methoden der Organischen Chemie; Schaumann, E., Ed.; Thieme: Stuttgart, 1994; Vol. E8c, pp 526–647; (g) Weaver, G. W. In Science of Synthesis: Houben-Weyl Methods of Molecular Transformations; Storr, R. C., Gilchrist, T. L., Eds.; Thieme: Stuttgart, 2003; Vol. 13, pp 219–251.

- 9. For example, see (a) Omar, F. A.; Mahfouz, N. M.; Rahman, M. A. Eur. J. Med. Chem. 1996, 31, 819–825; (b) Wang, X.; Li, Z.; Zhang, Z.; Da, Y. Synth. Commun. 2001, 31, 1907–1911.
- 10. (a) Batey, R. A.; Powell, D. A. Org. Lett. 2000, 2, 3237–3240; (b) Evindar, G.; Batey, R. A. Org. Lett. 2003, 5, 1201–1204.
- 11. Confalone, P. N.; Woodward, R. B. J. Am. Chem. Soc. 1983, 105, 902–906.
- 12. The Boc-thiosemicarbazides 2 did not require purification for use in the next step.
- 13. Hydrazide 1 (7.20 mmol) and arylisothiocyanate (7.20 mmol, 1.0 equiv) in CH_2Cl_2 (5 ml) were stirred for 20 min at room temperature. The solvent was removed in vacuo to give 2 which was immediately used in the next step. To a solution of 2 (0.36 mmol) in CH₃CN (3 ml) was added HgCl₂ (108.6 mg, 0.4 mmol) and triethylamine (0.088 ml, 1.10 mmol). The reaction mixture was stirred at room temperature for 12 h. The reaction slurry was filtered through a short plug of Celite and the solvent was removed in vacuo. The crude products were purified using semi-automated flash chromatography (silica gel, EtOAc/ hexanes 1:4) on a Biotage SP1 instrument.
- 14. Representative chararacterization data. (S) Isomer: {1-[5-(2-Bromophenylamino)-[1,3,4]oxadiazol-2-yl]-2-phenylethyl}-carbamic acid *tert*-butyl ester
4Ce: white solid; mp = 147–148 °C; [x] $^{26}_{10}$ – 53.0 (MeOH, *c 0.99); R_f* = 0.27 (EtOAc–hexanes; 1:4); IR (thin film) v 3350, 2949, 1700, 1696, 1590, 1521, 1450, 1376, 1308, 1276, 1206, 1009, 878, 720, 670 cm⁻¹; ¹H NMR (300 MHz,

CD₃OD–CD₂Cl₂ 3:1) δ 8.00 (1H, dd, J = 8.0, 1.0 Hz), 7.60 (1H, dd, J = 8.0, 1.0 Hz), 7.36 (1H, ddd, J = 8.0, 8.0, 1.2 Hz), 7.22-7.31 (5H, m), 7.02 (1H, ddd, J = 8.0, 8.0, 1.0 Hz), 5.02–5.09 (1H, m), 3.28 (1H, dd, J = 13.5, 6.5 Hz), 3.13 (1H, dd, J = 13.5,
9.0 Hz), 1.36 (9H, br s); ¹³C NMR (100 MHz, CD₃OD:CD₂Cl₂ 3:1) *δ* 161.6, 159.4, 155.2, 135.4, 133.4, 132.6, 128.9, 128.3, 127.1, 124.0, 120.9, 118.7, 111.6, 80.5, 42.6, 28.5, 22.8; HRMS (ESI) m/e ([M+1]⁺) calcd for C₂₁H₂₄N₄O₃Br 459.1024 found 459.1026.

- 15. Crystallographic data (excluding structure factors) for the structure of compound 4Ce have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 683066.
- 16. Purities of 5 and $\hat{6}$ were determined by HPLC/LCMS with the UV absorption measured at 254 nm.

18. Polymer-supported CMPI equivalent 8 is prepared by treatment of Wang resin with excess of 2-chloropyridine and triflic anhydride, see: (a) Crosignani, S.; Gonzalez, J.; Swinnen, D. Org. Lett. 2004, 6, 4579–4582; (b) Crosignani, S.; Swinnen, D. J. Comb. Chem. 2005, 7, 688–696.